

# Al-Mg-Sm (Aluminum-Magnesium-Samarium)

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The previous study of this system by [1987Zhe] presented an isothermal section at 400 °C, which depicts no ternary compounds. Recently, [2008Jia] carried out a thermodynamic assessment of this system and computed the isothermal section at 400 °C and a tentative liquidus projection.

## Binary Systems

The Al-Mg phase diagram [1998Lia] has the following intermediate phases: Mg<sub>2</sub>Al<sub>3</sub> (cubic, labeled β), R or ε (rhombohedral), and Mg<sub>17</sub>Al<sub>12</sub> (A12, αMn-type cubic, denoted γ). The Al-Sm system [1989Gsc, 2007Del] depicts the following intermediate phases: Sm<sub>3</sub>Al<sub>11</sub> (D1<sub>3</sub>, Al-deficient Al<sub>4</sub>Ba-type tetragonal), SmAl<sub>3</sub> (D0<sub>19</sub>, Ni<sub>3</sub>Sn-type hexagonal), SmAl<sub>2</sub> (C15, MgCu<sub>2</sub>-type cubic), SmAl (ErAl-type orthorhombic) and Sm<sub>2</sub>Al (C23, Co<sub>2</sub>Si-type orthorhombic). The Mg-Sm phase diagram [Massalski2] depicts the following compounds: Mg<sub>41</sub>Sm<sub>5</sub> (tetragonal), Mg<sub>5</sub>Sm (cubic), Mg<sub>3</sub>Sm (D0<sub>3</sub>, BiF<sub>3</sub>-type cubic), Mg<sub>2</sub>Sm (C15, MgCu<sub>2</sub>-type cubic), and MgSm (B2, CsCl-type

cubic). The computed phase diagrams of the above binaries were given by [2008Jia].

## Computed Ternary Phase Equilibria

The optimized binary interaction parameters of the Al-Sm and Mg-Sm systems were listed by [2008Jia] along with those adopted from the literature. As there are no ternary compounds in this system, [2008Jia] calculated the ternary phase equilibria by extrapolation of the binary data without introducing any ternary interaction parameters. The computed isothermal section at 400 °C shown in Fig. 1 was found to agree well with that determined by [1987Zhe]. The liquidus projection computed by [2008Jia] is shown in Fig. 2. In the binary Al-Sm phase diagram updated by [2007Del], there is only one modification of Sm<sub>3</sub>Al<sub>11</sub> (Al-deficient Al<sub>4</sub>Ba-type). Accordingly, the ternary peritectic reaction P<sub>2</sub>: L + SmAl<sub>3</sub> + Sm<sub>3</sub>Al<sub>11</sub>(HT) → Sm<sub>3</sub>Al<sub>11</sub>(LT) computed by [2008Jia] is omitted in Fig. 2. The reactions close to the Al-Mg side are shown schematically in the lower part of Fig. 2. An enlarged view of the Mg corner is

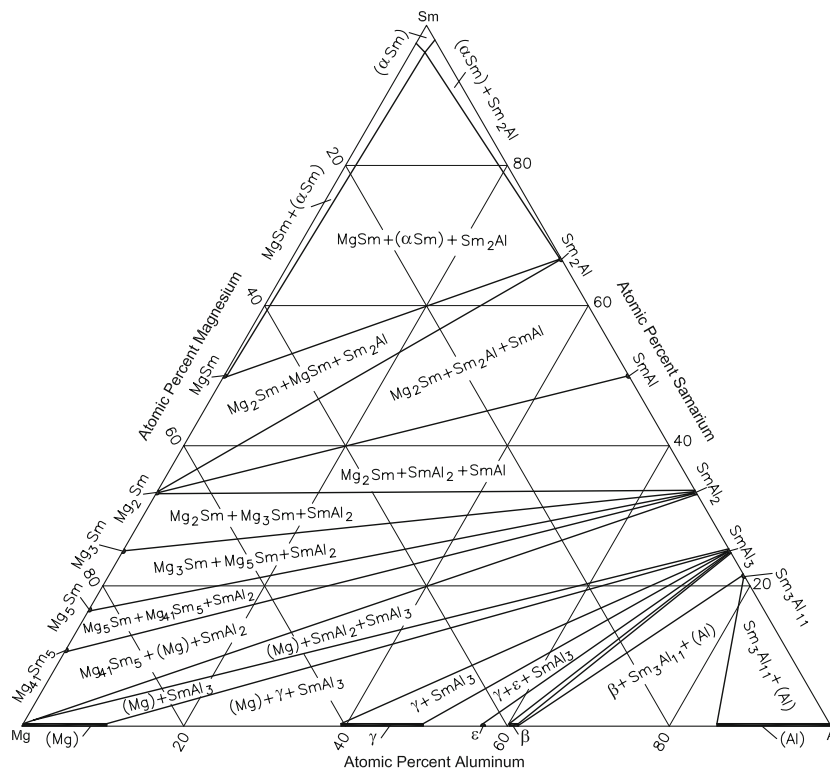
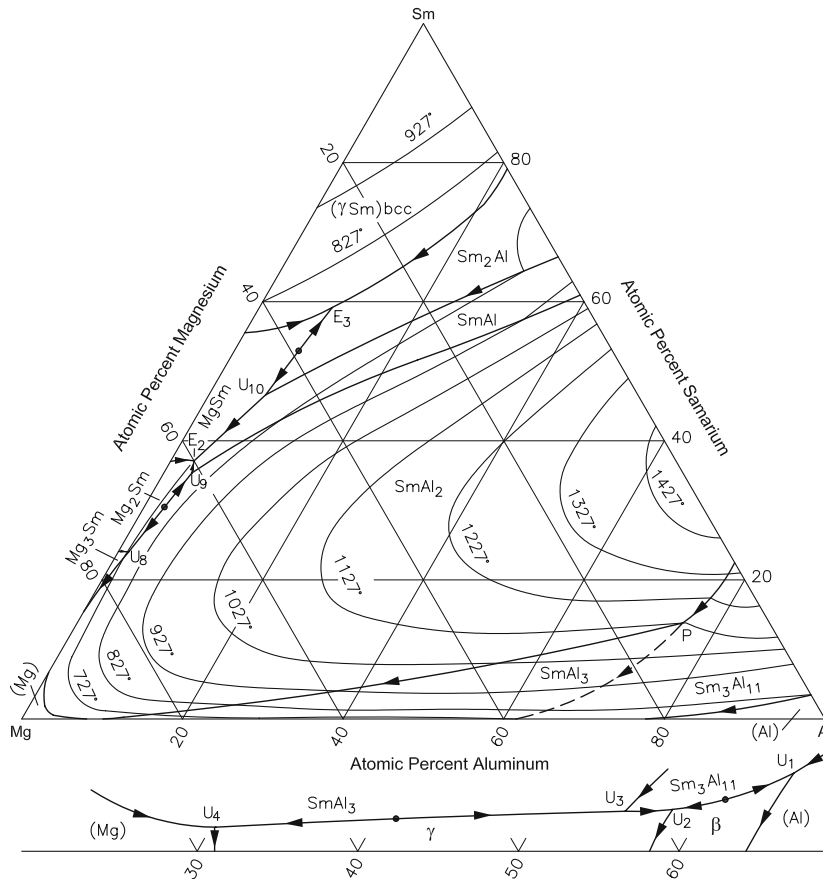
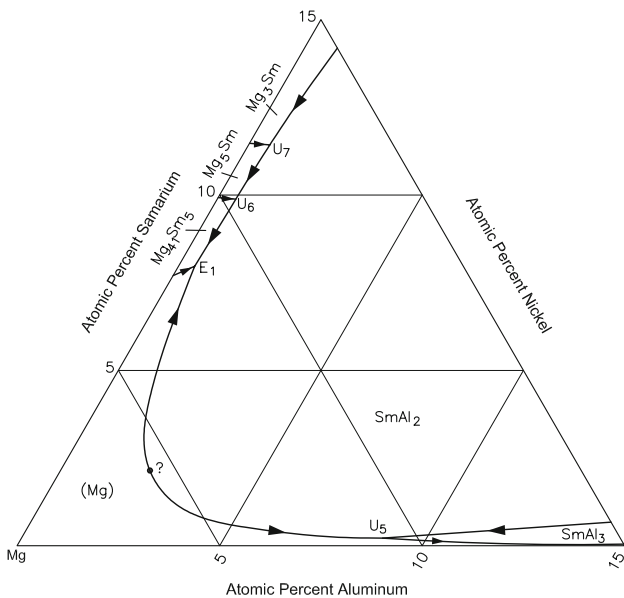


Fig. 1 Al-Mg-Sm computed isothermal section at 400 °C [2008Jia]

## Section II: Phase Diagram Evaluations



**Fig. 2** Al-Mg-Sm computed liquidus projection [2008Jia]



**Fig. 3** Al-Mg-Sm computed liquidus projection near the Mg corner [2008Jia]

shown in Fig. 3. The liquidus projection may be considered tentative, as there are no experimental data for comparison.

### References

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